AC 14/7/2016, Item No. 4.64

UNIVERSITY OF MUMBAI



Bachelor of Engineering

<u>First Year Engineering (Semester I & II), Revised course</u>

(REV- 2016) from Academic Year 2016 – 17,

(Common for All Branches of Engineering)

(As per Choice Based Credit and Grading System with effect from the academic year 2016–2017)

From Coordinator's Desk:-

To meet the challenge of ensuring excellence in engineering education, the issue of quality needs to be addressed, debated taken forward in a systematic manner. Accreditation is the principal means of quality assurance in higher education. The major emphasis of accreditation process is to measure the outcomes of the program that is being accredited. In line with this Faculty of Technology of University of Mumbai has taken a lead in incorporating philosophy of outcome based education in the process of curriculum development.

Faculty of Technology, University of Mumbai, in one of its meeting unanimously resolved that, each Board of Studies shall prepare some Program Educational Objectives (PEO's) give freedom to affiliated Institutes to add few (PEO's) course objectives course outcomes to be clearly defined for each course, so that all faculty members in affiliated institutes understand the depth approach of course to be taught, which will enhance learner's learning process. It was also resolved that, maximum senior faculty from colleges experts from industry to be involved while revising the curriculum. I am happy to state that, each Board of studies has adhered to the resolutions passed by Faculty of Technology, developed curriculum accordingly. In addition to outcome based education, **Choice Based Credit and Grading System** is also introduced to ensure quality of engineering education.

Choice Based Credit and Grading System enables a much-required shift in focus from teacher-centric to learner-centric education since the workload estimated is based on the investment of time in learning not in teaching. It also focuses on continuous evaluation which will enhance the quality of education. University of Mumbai has taken a lead in implementing the system through its affiliated Institutes Faculty of Technology has devised a transparent credit assignment policy adopted ten points scale to grade learner's performance. Credit grading based system was implemented for First Year of Engineering from the academic year 2016-2017. Subsequently this system will be carried forward for Second Year Engineering in the academic year 2017-2018, for Third Year Final Year Engineering in the academic years 2018-2019, 2019-2020, respectively.

Dr. S. K. Ukarande Co-ordinator, Faculty of Technology, Member - Academic Council University of Mumbai, Mumbai

Semester II

Course Code	Course Name		ching S ontact H			:			Cre	edits As	ssigned	l
Code		Theory	Prac	et.	Tu	ıt.	The	eory	TV	V/Pract	Tut.	Total
FEC201	Applied Mathematics-II	04	-		01	1	0	4		-	01	05
FEC202	Applied Physics-II	03	01		-		0	3		0.5	-	3.5
FEC203	Applied Chemistry -II	03	01		-		0	3		0.5	-	3.5
FEC204	Engineering Drawing	03	04		-		0	3		02	-	05
FEC205	Structured Programming Approach	04	02		-		0	4		01	-	05
FEC206	Communication Skills	02	02		-		0	2		01	-	03
FEL201	Basic Workshop Practice-II	-	04		-		-	-		02	-	02
Total	Total			1				9		07	01	27
						kami	inat	ion S	Sch	eme	T	
Course	Course Name	Theory Internal Assessment										
Code		Intern	ai Asses	sme			nd	Ter		Pract	Oral	Total
		Test1	Test2	A	vg		Sem Work		rk	Truct	Orai	Total
FEC201	Applied Mathematics-II	20	20	2	0	80	0	25		-	-	125
FEC202	Applied Physics-II	15	15	1	5	60	0	25		-	-	100
FEC203	Applied Chemistry -II	15	15	1	5	60	0	25	;	-	-	100
FEC204	Engineering Drawing	15	15	1	5	60	0	25	i	50	-	150
FEC205	Structured Programming Approach	20	20	2	0	80	0	25	į	25	-	150
FEC206	Communication Skills	10	10	1	0	40	0	25		-	-	75
FEL201	Basic Workshop Practice-II	-	1	-	-	-		50)	-	-	50
Total				9	5	38	80	200	0	75	-	750

Course Code	Course Name		hing Scho ntact Hou		Credits Assigned				
Code		Theory	Pract.	Tut.	Theory	TW/Pract	Tut.	Total	
FEC201	Applied Mathematics-II	04		01	04		01	05	

Course	Course Name	Examination Scheme									
			Tl	heory							
		Internal Assessment				Term					
Code		Test1	Test2	Av of Test 1 & 2	End Sem Exam	Work	Pract	Oral	Total		
FEC201	Applied Mathematics-II	20	20	20	80	25			125		

- 1. To provide students with sound foundation in applied mathematics to solve real life problems in industry.
- 2. To provide hands on experience in using Scilab software to handle real life problems.

- 1. Apply the concepts of First Order and first degree Differential equation to the engineering problems.
- 2. Apply the concepts of Higher Order Linear Differential equation to the engineering problems.
- 3. Apply concepts of Beta and Gamma function to the engineering Problems.
- 4. Apply SCILAB programming techniques to solve differential equation to model complex engineering activities.
- 5. Apply concepts of Double integral of different coordinate systems to the engineering problems.
- 6. Apply concepts of triple integral of different coordinate systems to the engineering problems.

Module	Detailed Contents	Hrs.
	Differential Equations of First Order and First Degree	
	1.1 Exact differential Equations, Equations reducible to exact form by using integrating	
	factors.	4
01	1.2 Linear differential equations (Review), equation reducible to linear form, Bernoulli's	
	equation.	3
	1.3: Simple application of differential equation of first order and first degree to electrical	
	and Mechanical Engineering problem (no formulation of differential equation)	2
	Linear Differential Equations With Constant Coefficients and Variable Coefficients	
02	Of Higher Order	
	2.1. Linear Differential Equation with constant coefficient- complementary function,	6
	particular integrals of differential equation of the type $f(D)y = X$ where X is e^{ax} , $\sin(ax+b)$,	
	$\cos(ax+b), x^n, e^{ax}V, xV.$	
	2.2. Cauchy's homogeneous linear differential equation and Legendre's differential	3
	equation, Method of variation of parameters.	
	Numerical solution of ordinary differential equations of first order and first degree,	
	Beta and Gamma Function	_
03	3.1. (a)Taylor's series method (b)Euler's method	4
	(c) Modified Euler method (d) Runga-Kutta fourth order formula (SciLab programming is	
	to be taught during lecture hours)	
	3.2 .Beta and Gamma functions and its properties.	4
	Differentiation under Integral sign, Numerical Integration and Rectification	
0.4	4.1. Differentiation under integral sign with constant limits of integration.	2
04	4.2. Numerical integration- by (a) Trapezoidal (b) Simpson's 1/3rd (c) Simpson's 3/8th rule	3
	(all with proof). (Scilab programming on (a) (b) (c) (d) is to be taught during lecture hours)	2
	4.3. Rectification of plane curves.	3

05	 Double Integration 5.1. Double integration-definition, Evaluation of Double Integrals. 5.2. Change the order of integration, Evaluation of double integrals by changing the order of integration and changing to polar form. 	2 7
06	 Triple Integration and Applications of Multiple Integrals. 6.1. Triple integration definition and evaluation (Cartesian, cylindrical and spherical polar coordinates). 6.2. Application of double integrals to compute Area, Mass, Volume. Application of triple integral to compute volume. 	3

Term Work:

General Instructions:

- 1. Batch wise tutorials are to be conducted. The number of students per batch should be as per University pattern for practical.
- 2. Students must be encouraged to write Scilab Programs in tutorial class only. Each Student to write atleast 4 Scilab tutorials (including print out) and at least 6 class tutorials on entire syllabus.
- 3. SciLab Tutorials will be based on (i)Curve Tracing (ii) Taylor's series method, Euler's method Modified Euler method, RungaKutta fourth order formula (iii) Ordinary Differential Equation and (iv) Trapezoidal Simpson's 1/3rd and Simpson's 3/8th rule.

The distribution of Term Work marks will be as follows -

Attendance (Theory and Tutorial): 05 marks Class Tutorials on entire Syllabus: 10 marks SciLab Tutorials : 10 marks

Assessment:

Internal Assessment Test:

Assessment consists of two class tests of 20 marks each. The first class test is to be conducted when approx. 40% syllabus is completed and second class test when additional 35% syllabus is completed. Duration of each test shall be one hour.

End Semester Theory Examination:

- 1. Question paper will comprise of total 06 questions, each carrying 20 marks.
- 2. Total 04 questions need to be solved.
- 3. Question No: 01 will be compulsory and based on entire syllabus wherein sub-questions of 3 to 4 marks will be asked.
- 4. Remaining questions will be mixed in nature.(e.g. Suppose Q.2 has part (a) from module 3 then part (b) will be from any module other than module 3)
- 5. In question paper weightage of each module will be proportional to number of respective lecture hrs as mentioned in the syllabus.

- 1. A text book of Applied Mathematics, P.N.Wartikar and J.N.Wartikar, Vol I and –II by Pune VidyarthiGraha.
- 2. Higher Engineering Mathematics, Dr.B.S.Grewal, Khanna Publication
- 3. Advanced Engineering Mathematics, Erwin Kreyszig, Wiley EasternLimited, 9thEd.
- 4. Numerical methods by Dr. P. Kandasamy ,S. Chand Publications

Course Code	Course Name		hing Sche tact Hou		Credits Assigned				
Code		Theory	Pract.	Tut.	Theory	TW/Pract	Tut.	Total	
FEC202	Applied Physics – II	03	01		03	0.5		3.5	

	Course Name	Examination Scheme									
Course			Tl	heory							
		Inter	nal Asse	essment	End	Term					
Code		Test1	Test2	Av of Test 1 & 2	Sem Exam Work		Pract	Oral	Total		
FEC202	Applied Physics – II	15	15	15	60	25			100		

- 1. To impart knowledge of basic concepts in applied physics.
- 2. To provide the knowledge and methodology necessary for solving problems in the field of engineering.

- 1. Comprehend principles of interference and diffraction.
- 2. Illustrate the principle, construction and working of various LASERs and its applications.
- 3. Identify various applications of optical fibres.
- 4. Comprehend the concepts of electrodynamics and Maxwell's equations and their use in telecommunication systems.
- 5. Apply the concepts of electromagnetism in focusing systems and CRO.
- 6. Comprehend the significance of nanoscience and nanotechnology, its applications.

Module	Detailed Contents	Hrs.
01	Interference by division of amplitude and by division of wave front; Interference in thin film of constant thickness due to reflected and transmitted light; origin of colours in thin film; Wedge shaped film(angle of wedge and thickness measurement); Newton's rings Applications of interference - Determination of thickness of very thin wire or foil; determination of refractive index of liquid; wavelength of incident light; radius of curvature of lens; testing of surface flatness; Anti-reflecting films and Highly reflecting film. Diffraction of Light –Fraunhoffer diffraction at single slit, Fraunhoffer diffraction at double slit, Diffraction Grating, Resolving power of a grating, dispersive power of a grating Application of Diffraction - Determination of wavelength of light with a plane transmission grating	14
02	LASERS Quantum processes as absorption, spontaneous emission and stimulated emission; metastable states, population inversion, pumping, resonance cavity, Einsteins's equations; Helium Neon laser; Nd:YAG laser; Semiconductor laser, Applications of laser- Holography (construction and reconstruction of holograms) and industrial applications(cutting, welding etc), Applications in medical field	04
03	FIBRE OPTICS Total internal reflection; Numerical Aperture; critical angle; angle of acceptance; Vnumber; number of modes of propagation; types of optical fiber; Losses in optical fibre(Attenuation and dispersion) Applications of optical fibre - Fibre optic communication system; sensors (Pressure, temperature, smoke, water level), applications in medical field	04

04	ELECTRODYNAMICS Cartesian, Cylindrical and Spherical Coordinate system, Scaler and Vector field, Physical significance of gradient, curl and divergence, Determination of Maxwell's four equations. Applications-design of antenna, wave guide, satellite communication etc.	08
05	CHARGE PARTICLE IN ELECTRIC AND MAGNETIC FIELDS Fundamentals of Electromagnetism, Motion of electron in electric field (parallel ,perpendicular, with some angle); Motion of electron in magnetic field (Longitudinal and Transverse); Magnetic deflection; Motion of electron in crossed field; Velocity Selector; Velocity Filter, Electron refraction; Bethe's law; Electrostatic focusing; Magnetostatic focusing; Cathode ray tube (CRT); Cathod ray Oscilloscope (CRO) Application of CRO: Voltage (dc,ac), frequency, phase measurement.	05
06	Introduction to nano-science and nanotechnology, Surface to volume ratio, Two main approaches in nanotechnology -Bottom up technique and top down technique; Important tools in nanotechnology such as Scanning Electron Microscope, Transmission Electron Microscope, Atomic Force Microscope. Nano materials: Methods to synthesize nanomaterials (Ball milling, Sputtering, Vapour deposition, solgel), properties and applications of nanomaterials.	04

Suggested Experiments: (Any five)

- 1. Determination of radius of curvature of a lens using Newton's ring set up
- 2. Determination of diameter of wire/hair or thickness of paper using Wedge shape film method.
- 3. Determination of wavelength using Diffracion grating. (Hg/ Ne source)
- 4. Determination of number of lines on the grating surface using LASER Sourse.
- 5. Determination of Numerical Aperture of an optical fibre.
- 6. Determination of wavelength using Diffracion grating. (Laser source)
- 7. Use of CRO for measurement of frequency and amplitude.
- 8. Use of CRO for measurement of phase angle.
- 9. Study of divergence of laser beam
- 10. Determination of width of a slit using single slit diffraction experiment (laser source)

The distribution of Term Work marks will be as follows –

4. Attendance (Theory and Practical) : 05 marks
5. Assignments : 10 marks
6. Laboratory work (Experiments and Journal) : 10 marks

Assessment:

Internal Assessment Test:

Assessment consists of two class tests of 15 marks each. The first class test is to be conducted when approx. 40% syllabus is completed and second class test when additional 35% syllabus is completed. Duration of each test shall be one hour.

End Semester Theory Examination:

- 1. Question paper will comprise of total 06 questions, each carrying 15 marks.
- 2. Total 04 questions need to be solved.
- 3. Question No: 01 will be compulsory and based on entire syllabus wherein sub-questions of 2 to 3marks will be asked.
- 4. Remaining questions will be mixed in nature.(e.g. Suppose Q.2 has part (a) from module 3 then part (b) will be from any module other than module 3)
- 5. In question paper weightage of each module will be proportional to number of respective lecture hrs as mentioned in the syllabus.

- 1. A text book of Engineering Physics-Avadhanulu&Kshirsagar, S.Chand
- 2. Fundamentals of Optics by Jenkins and White, McGraw-Hill
- 3. Optics Ajay Ghatak, Tata McGraw Hill
- 4. Concepts of Modern Physics- ArtherBeiser, Tata Mcgraw Hill
- 5. A textbook of Optics N. Subramanyam and Brijlal, S.Chand
- 6. Engineering Physics-D. K. Bhattacharya, Oxford
- 7. Concepts of Modern Physics- ArtherBeiser, Tata Mcgraw Hill
- 8. Classical Electodyamics J. D. Jackson, Wiley
- 9. Introduction to Electrodynamics- D. J. Griffiths, Pearson publication
- 10. Intoduction to Nanotechnology- Charles P. Poole, Jr., Frank J. Owens, Wiley India edition
- 11. Nano: The Essential T. Pradeep, Mcgraw-Hill Education

Course Code	Course Name		hing Scho ntact Hou		Credits Assigned				
Code		Theory	Pract.	Tut.	Theory	TW/Pract	Tut.	Total	
FEC203	Applied Chemistry – II	03	01		03	0.5		3.5	

		Examination Scheme										
Course			T	heory								
	Course Name	Internal Assessment				Term						
Code	Course Mane	Test1	Test2	Av of Test 1 & 2	End Sem Exam	Work Pract	Oral	Total				
FEC203	Applied Chemistry – II	15	15	15	60	25			100			

- 1. To provide necessary background in applied chemistry relevant to chemical industries.
- 2. To provide exposure in conducting experiments and interpret and report the results in professional format.

- 1. Identify types of corrosion and factors affecting it related to problems affecting all industries.
- 2. Identify different types of corrosion control methods to study corrosion control in various industries.
- 3. Apply the knowledge of different types of fuels, including their production and refining methods and combustion mechanisms.
- 4. Illustrate composition and properties of different types of alloys and the process of powder metallurgy
- 5. Illustrate principales of green chemistry.
- 6. Illustrate properties and applications of different types of composite materials.

Module	Detailed Contents	Hrs.
01	Corrosion: Introduction: Types of Corrosion- (I) Dry or Chemical Corrosion-i) Due to oxygen ii) Due to other gases (II) Wet or Electrochemical corrosion- Mechanism i) Evolution of hydrogen type ii) Absorption of oxygen. Types of Electrochemical Corrosion- Galvanic cell corrosion, Concentration cell corrosion (differential aeration), Pitting corrosion, Intergranular corrosion, Stress corrosion. Factors affecting the rate of corrosion- Nature of metal, position of metal in galvanic series, potential difference, overvoltage, relative area of anodic and cathodic parts, purity of metal, nature of the corrosion product, temperature, moisture, influence of pH, concentration of the electrolytes. Methods to decrease the rate of corrosion-Material selection, Proper designing, Use of inhibitors, Cathodic protection- i) Sacrificial anodic protection ii) Impressed current method, Anodic protection method, Metallic coatings- hot dipping- galvanizing and tinning, metal cladding, metal spraying, Electroplating, Cementation. Organic coatings — Paints (only constituents and their functions).	11
02	Alloys Introduction, purpose of making alloys, Ferrous alloys, plain carbon steel, heat resisting steels, stainless steels (corrosion resistant steels), effect of the alloying element- Ni, Cr, Co, Mn, Mo,W and V; Non-Ferrous alloys- Composition, properties and uses of- Alloys of Aluminium- i) Duralumin ii) Magnalium. Alloys of Cu- (I) Brasses-i) Commercial brass ii) German silver, (II) Bronzes- i) Gun metal ii) High phosphorous bronze. Alloys of Pb- i) Wood's metal ii)	07

	Tinmann's solder. Powder Metallurgy- Introduction, (1)Methods of powder metal	
	formation- i) Mechanical pulverization ii) Atomization iii) Chemical reduction iv)	
	Electrolytic process v) Decomposition (2) Mixing and blending. (3) Sintering (4)	
	Compacting- i) Cold pressing ii) Powder injection moulding (iii) Hot compaction.	
	Applications of powder metallurgy.	
	Shape Memory Alloys- Definition, properties and Uses.	
	Fuels	
	Definition, classification of fuels-solid, liquid and gaseous. Calorific value- Definition,	
	Gross or Higher calorific value & Net or lower calorific value, units of heat (no conversions),	
	Dulong's formula & numerical for calculations of Gross and Net calorific values.	
	Characteristics of a good fuel.	
	Solid fuels- Analysis of coal- Proximate and Ultimate Analysis with Significance and	
	numericals.	
	Liquid fuels- Crude petroleum oil, its composition and classification and mining (in brief).	
	Refining of crude oil- i) Separation of water ii) Separation of 'S' & iii) Fractional Distillation	
	with diagram and composition and uses table.	
03	Cracking- Definition, Types of cracking-	12
	I) Thermal cracking – (i) Liquid phase thermal cracking (ii) Vapour phase thermal cracking.	
	II) Catalytic cracking- (i) Fixed-bed catalytic cracking (ii) Moving-bed catalytic cracking.	
	Advantages of Catalytic cracking.	
	Petrol- Refining of petrol, unleaded petrol (use of MTBE), Catalytic converter, Power	
	alcohol, Knocking, Octane number, Cetane number, Antiknocking agents.	
	Combustion- Calculations for requirement of only oxygen and air (by weight and by volume	
	only) for given solid & gaseous fuels.	
	Biodiesel- Method to obtain Biodiesel from vegetable oils (Trans-esterification), advantage	
	and disadvantages of biodiesel.	
	Fuel cell- Definition, types and applications.	
	Composite Materials	
	Introduction, Constitution- i) Matrix phase ii) Dispersed phase. Characteristic properties of	
	composite materials. Classification- (A) Particle - reinforced composites- i) Large – particle	
04	reinforced composites ii) Dispersion – strengthened composites. (B) Fiber – reinforced	04
	composites- i) Continuous – aligned ii) Discontinuous – aligned (short)- (a) aligned (b)	
	randomly oriented (C) Structural Composites- i) Laminates (ii) Sandwich Panels.	
	Green Chemistry	
05	Introduction, Twelve Principles of Green chemistry, numerical on atom economy,	06
	Conventional and green synthesis of Adipic acid, Indigo, Ibuprofen and Carbaryl.	
	Green solvents (water, supercritical CO ₂) and products from natural materials.	

Suggested Experiments: (Any five)

- 1. Estimation of Zn- Complexometric titration.
- 2. Estimation of Ni- Complexometric titration.
- 3. Estimation of Al- Complexometic titration.
- 4. Flue gas analysis using Orsat's apparatus.
- 5. Estimation of Fe from plain carbon steel
- 6. Estimation of Ni by gravimetric method.
- 7. Estimation of Sniodometrically.
- 8. Preparation of Biodiesel from edible oil.
- 9. Estimation of Cu- Iodometrically.
- 10. Estimation of percentage moisture in coal.
- 11. Estimation of percentage ash in coal.
- 12. To estimate the emf of Cu-Zn system by potentiometry.
- 13. Demonstration of Electroplating.

Term work

Term Work shall consist of minimum five experiments. The distribution of marks for term work shall be as follows:

Attendance (Practical and Theory) : 05 marks
 Laboratory Work (Experiments and journal) : 10 marks
 Assignments and Viva on practicals : 10 marks

Assessment:

Internal Assessment Test:

Assessment consists of two class tests of 15 marks each. The first class test is to be conducted when approx. 40% syllabus is completed and second class test when additional 35% syllabus is completed. Duration of each test shall be one hour.

End Semester Theory Examination:

- 1. Question paper will comprise of total 06 questions, each carrying 15 marks.
- 2. Total 04 questions need to be solved.
- 3. Question No: 01 will be compulsory and based on entire syllabus wherein sub-questions of 3 marks will be asked.
- 4. Remaining questions will be mixed in nature.(e.g. Suppose Q.2 has part (a) from module 3 then part (b) will be from any module other than module 3)
- 5. In question paper weightage of each module will be proportional to number of respective lecture hrs as mentioned in the syllabus.

- 1. Engineering Chemistry Jain & Jain (DhanpatRai)
- 2. Engineering Chemistry Dara & Dara (S Chand)
- 3. Engineering Chemistry Wiley India (ISBN 9788126519880)
- 4. A Text Book of Engineering Chemistry ShashiChawla (DhanpatRai)
- 5. A Text Book of Green Chemistry V.K. Ahluwalia (Springer)

Course Code	Course Name		hing Scho ntact Hou			Credits Ass	signed	
Code		Theory	Pract.	Tut.	Theory	TW/Pract	Tut.	Total
FEC204	Engineering Drawing	03	04		03	02		05

				Exa	minatior	n Scheme					
			T	heory							
Course	Course Name	Inter	nal Asso	essment	End	Term		Oral			
Code	Course rune	Test1	Test2	Av of Test 1 & 2	Sem Exam	Work	Pract	Oral	Total		
FEC204	Engineering Drawing	15	15	15	60	25	50		150		

- 1. To impart and inculcate proper understanding of the theory of projection.
- 2. To impart the knowledge of reading a drawing.
- 3. To improve the visualization skill.
- 4. To teach basic utility of computer aided drafting (CAD) tool.

- 1. Apply the basic principles of projections in 2D drawings.
- 2. Apply the basic principles of projections in converting 3D view to 2D drawing.
- 3. Read a given drawing.
- 4. Visualize an object from the given two views.
- 5. Use CAD tool to draw different views of a 3D object.
- 6. Use CAD tool to draw an object in 3D.

Module	Detailed Contents	Hrs.
	Introduction to Engineering Drawing:- Types of Lines, Dimensioning Systems as per	
	IS conventions.	
01	Engineering Curves: Basic construction of Cycloid, Involutes and Helix (of cylinder)	
VI	only.	3
	** Introduction to Auto CAD:- Basic Drawing and Editing Commands. Knowledge of	
	setting up layers, Dimensioning, Hatching, plotting and Printing.	
	<u>Projection of Points and Lines:</u> Lines inclined to both the Reference Planes	
	(Excluding Traces of lines) and simple application based problems on Projection of	
02	lines.	6
	@Projection of Planes:- Triangular, Square, Rectangular, Pentagonal, Hexagonal and	
	Circular planes inclined to either HP or VP only. (Exclude composite planes)	
03	Projection of Solids:- (Prism, Pyramid, Cylinder, Tetrahedron, Hexahedron and Cone only) Solid projection with the axis inclined to HP and VP. (Exclude Spheres, Composite, Hollow solids and frustum of solids). Use change of position or Auxiliary plane method Section of Solids:- Section of Prism, Pyramid, Cylinder, Tetrahedron, Hexahedron & Cone cut by plane perpendicular to at least one reference plane. (Exclude Curved Section Plane). Use change of position or Auxiliary plane method Development of Lateral Surfaces of Sectioned Solids:- Lateral surface development of Prism, Pyramid, Tetrahedron, Hexahedron, Cylinder, Cone with section plane inclined to HP or VP only. (Exclude DLS of a solid with a hole in it and Reverse Development). (Exclude Reverse Development)	14
04	 Orthographic and Sectional Orthographic Projections:- Different views of a simple machine part as per the first angle projection method recommended by I.S. Full or Half Sectional views of the Simple Machine parts. **Drawing of orthographic projections using Auto CAD. 	12

05	Isometric Views:- Isometric View/Drawing of blocks of plain and cylindrical surfaces using plain/natural scale only. (Exclude Spherical surfaces). **Prawing of Isometric views using Auto CAD. *@Reading of Orthographic Projections. [Only for Practical Exam (AutoCAD) and Term Work] **Orthographic Reading using Auto CAD. **Introduction to 3D in AutoCAD Working in 2 dimensions Viewing 2D Objects, Pagin wineframe models, Extrading simple.	10

^{**}Should be covered during Auto CAD practical sessions.

TERM WORK:

Component – 1

Drawing Sheet – 1: Projection of Solids (3 Problems)

Drawing Sheet – 2: Section of Solids and Development of lateral surfaces (2 Problems)

Drawing Sheet – 3: Orthographic Projection without section (2 Problems)

Drawing Sheet – 4: Orthographic Projection with section (2 Problems)

Drawing Sheet – 5: Isometric Views (3 Problems)

Component -2

One A-3 size sketch book consisting of:-

- Two problems each from Engineering Curves, Projection of Lines, Planes and Solids.
 One problem from Section of solids without DLS and one problem from section of solids with DLS of that sectioned Solid.
- 2) Two problems from Orthographic Projections (with Section), One problem on Reading of Orthographic projections and Two problems on Isometric views.

Component-3

Printouts (preferably on A3 size sheet) of each from:

- 1. Orthographic Projections with Section 3 problems.
- 2. Isometric Views 4 problems
- 3. Reading of Orthographic Projections 1 problem.

Note:- 2 hrs /week Auto CAD Practical is essential for completing the Auto CAD Drawings and take required printouts.

AUTO CAD PRACTICAL EXAMINATION: (2hrs – 50 marks):

1) Minimum 1 problem from 1 <u>OR</u> 3 of Component-3 <u>for 30 marks</u>.

(All three views with at least 12 dimensions must be asked in the exam)

AND

2) Minimum 1 problem from 2 of Component-3 for 20 marks.

Note:- Print out of the Answers have to be taken **preferably in A3 size sheets** and should be **Assessed by External examiner only**. Knowledge of concepts and accuracy of drawing should be considered during evaluation.

[@] Should be covered only in Term work. (i.e. Questions will not be asked for the End semester Examination).

INTERNAL ASSESSMENT TEST: (1 hr - 15 marks)

Out of the two tests, one test must be conducted by **conventional way** and another test must be **Practical Exam** (using AutoCAD software). Average of the two tests must be considered for Internal Assessment.

END SEMESTER EXAMINATION: (3 hrs – 60 marks)

- 1) Question paper will comprise of 6 questions, each carrying 15 marks.
- 2) Any 4 questions need to be solved. There won't be any compulsory Question.
- 3) Marks of each topic should be proportional to number of hours assigned to each Module.

Text Books.

- 1 N.D. Bhatt, "Engineering Drawing (Plane and solid geometry)", Charotar Publishing House Pvt. Ltd.
- 2 N.D. Bhatt & V.M. Panchal, "Machine Drawing", Charotar Publishing House Pvt. Ltd.

- 1 M.B Shah & B.C Rana, "Engineering Drawing", Pearson Publications.
- 2 P.J. Shah, "Engineering Graphics", S Chand Publications.
- 3 Dhananjay A Jolhe, "Engineering Drawing" Tata McGraw Hill.
- 4 Prof. Sham Tickoo (Purdue University) & Gaurav Verma, "(CAD Soft Technologies): Auto CAD 2012 (For engineers and Designers)", Dreamtech Press New Delhi.

Course Code	Course Name	(Contact Hours)					C reality Accidned			
Code		Theory	Pract.	Tut.	Theory	TW/Pract	Tut.	Total		
	Structured									
FEC205	Programming	04	02		04	01		05		
	Approach									

				Exa	minatior	Schem	e		
			Tl	heory					
Course	Course Name	Inter	nal Asso	essment	End	Term			
Code	Course Manie	Test1	Test2	Av of Test 1 & 2	Sem Exam	Work	Pract	Oral	Total
FEC205	Structured Programming Approach	20	20	20	80	25	25		150

- 1. To familirise the logic of structured programming approach.
- 2. To provide exposure in developing algorithm, flowchart and thereby writing efficient codes for user defined problem.

- 1. Illustrate the basic terminology used in computer programming.
- 2. Illustrate the concept of data types, variables and operators using C.
- 3. Design and Implement control statements and looping constructs in C.
- 4. Apply function concept on problem statements.
- 5. Demonstrate the use of arrays, strings, structures and files handling in C.
- 6. Demonstrate the dynamics of memory by the use of pointers to construct various data structures.

Module	Topic	Detailed Contents	Hrs.
01	Introduction to Computer, Algorithm And Flowchart	 1.1 Basics of Computer: Turing Model, Von Neumann Model, Basics of Positional Number System, Introduction to Operating System and component of an Operating System. 1.2 Algorithm & Flowchart: Three construct of Algorithm and flowchart: Sequence, Decision (Selection) and Repetition 	06
02	Fundamentals of C-Programming	 2.1 Character Set, Identifiers and keywords, Data types, Constants, Variables. 2.2 Operators-Arithmetic, Relational and logical, Assignment, Unary, Conditional, Bitwise, Comma, other operators. Expression, statements, Library Functions, Preprocessor. 2.3 Data Input and Output – getchar(), putchar(), scanf(), printf(), gets(), puts(), Structure of C program . 	06
03	Control Structures	 3.1 Branching - If statement, If-else Statement, Multiway decision. 3.2 Looping – while , do-while, for 3.3 Nested control structure- Switch statement, Continue statement Break statement, Goto statement. 	12
04	Functions and Parameter	 4.1Function -Introduction of Function, Function Main, Defining a Function, Accessing a Function, Function Prototype, Passing Arguments to a Function, Recursion. 4.2 Storage Classes –Auto, Extern, Static, Register 	06

05	Arrays , String Structure and Union	 5.1 Array-Concepts, Declaration, Definition, Accessing array element, One-dimensional and Multidimensional array. 5.2 String- Basic of String, Array of String, Functions in String.h 5.3 Structure- Declaration, Initialization, structure within structure, Operation on structures, Array of Structure. 5.4 Union - Definition, Difference between structure and union, Operations on a union 	14
06	Pointer and Files	 6.1 Pointer: Introduction, Definition and uses of Pointers, Address Operator, Pointer Variables, Dereferencing Pointer, Void Pointer, Pointer Arithmetic, Pointers to Pointers, Pointers and Array, Passing Arrays to Function, Pointers and Function, Pointers and two dimensional Array, Array of Pointers, Dynamic Memory Allocation. 6.2 Files: Types of File, File operation- Opening, Closing, Creating, Reading, Processing File. 	08

Laboratory Assignments:

- 1. Students are expected to solve and execute at least 20 programming problems based on above Syllabus.
- 2. Journal work should comprise of writing the problem definition, solution of problem either as algorithm and flow chart and source code in C (Advisable hand written) for all the 20 problems.

Assessment:

Internal Assessment Test:

Assessment consists of two class tests of 20 marks each. The first class test is to be conducted when approx. 40% syllabus is completed and second class test when additional 35% syllabus is completed. Duration of each test shall be one hour.

End Semester Theory Examination:

- 1. Question paper will comprise of total 06 questions, each carrying 20 marks.
- 2. Total 04 questions need to be solved.
- 3. Question No: 01 will be compulsory and based on entire syllabus wherein sub-questions of 3 to 5 marks will be asked.
- 4. Remaining questions will be mixed in nature.(e.g. Suppose Q.2 has part (a) from module 3 then part (b) will be from any module other than module 3)
- 5. In question paper weightage of each module will be proportional to number of respective lecture hrs as mentioned in the syllabus.

Text Books:

- 1 "MASTERING C" by K.R.Venugopal and SudeepR.Prasad, Tata McGraw-Hill Publications.
- 2 "A Computer Science Structure Programming Approaches using C", by BehrouzForouzan , Cengage Learning .
- 3 Schaum's outlines "Programming with C", by Byron S. Gottfried, Tata McGraw-Hill Publications.

Reference Books:

- 1 "Basics of Computer Science", by BehrouzForouzan, Cengage Learning.
- 2 "Programming Techniques through C", by M. G. Venkateshmurthy, Pearson Publication.
- 3 "Programming in ANSI C", by E. Balaguruswamy, Tata McGraw-Hill Education.
- 4 "Programming in C", by Pradeep Day and Manas Gosh, Oxford University Press.
- 5 "Let Us C", by YashwantKanetkar, BPB Publication.

Course Code	Course Name		hing Scho ntact Hou			Credits Assigned TW/Pract Tut. 01		
Code		Theory	Pract.	Tut.	Theory	TW/Pract	Tut.	Total
FEC206	Communication Skills	02	02		02	01		03

				Exa	minatior	Schem	ne				
			T	heory							
Course	Course Name	Inter	nal Asso	essment	End	Term					
Code	00 01 30 1 (01 10	Test1	Test2	Av of Test 1 & 2	Sem Exam	Work	Pract	Oral	Total		
FEC206	Communication Skills	10	10	10	40	25			75		

- 1. To acquaint the students with appropriate language skills with the purpose of improving the existing ones LSRW
- 2. To make the learners understand the importance and effective use of non-verbal communication
- 3. To make the learner proficient in public speaking and presentation skills
- 4. To guide and teach the students to utilize the principles of professional business and technical writing for effective communication in the global world
- 5. To make the learner capable of creating official content digitally for further communication in the corporate environment

- 1. Understand and evaluate information they listen to and express their ideas with greater clarity
- 2. Speak and respond effectively along the various channels of communication in a business organization
- 3. Speak convincingly before an audience with the help of an expanded vocabulary and enhanced digital content
- 4. Read and summarize effectively
- 5. Communicate through result oriented writing both within and outside the organization.
- 6. Write a set of effective and easy to understand technical description, instructions and convey the same using global information technology

Module	Detailed Contents	Hrs.
01	Communication Theory: Concept and Meaning, Communication cycle, Objectives, Barriers to communication (linguistic and semantic, psychological, physical, mechanical, cultural), Methods of communication (verbal and non-verbal), Networks of communication (formal and informal), Language skills (listening, speaking, reading, writing), Corporate communication: Digital Content Creation.	13
02	Business Correspondence: Principles of Business Correspondence, Parts of a business letter, Formats (Complete block and Modified block), Types of letters: Enquiry, Reply to enquiry, Claim, Adjustment and Sales letter.	05
03	Grammar and Vocabulary: Common errors, Concord (subject- verb agreement), Pairs of confused words, Lexicon (Enriching vocabulary through one-word substitutes, synonyms, antonyms, etc.)	02

04	Summarization and Comprehension: Passages to test the analytical skills and expression	02
05	Technical writing: Techniques to define an object, writing instructions, language exercises based on types of expositions (description of an object, explanation of a process)	02
06	Information Communication Technology (ICT) enabled communication media: E-mail, Blog and Website.	02

The distribution of Term Work marks will be as follows -

Attendance : 05 marks Assignments : 20 marks

List of assignments:

- 1. Communication theory: 02
- 2. Business Correspondence: 02
- 3. Grammar and vocabulary: 01
- 4. Summarization & Comprehension: 01
- 5. Technical writing: 01
- 6. ICT enabled communication media: 01

Assessment:

Internal Assessment Test:

Assessment consists of two class tests of 10 marks each. The first test should be conducted in the form of a three-minute public speech. The second test should be based on theory and application exercises as mentioned in the syllabus.

End Semester Theory Examination:

- 1. Question paper will comprise of total 06 questions, each carrying 10 marks.
- 2. Total 04 questions need to be solved.
- 3. Question No: 01 will be compulsory and based on entire syllabus wherein sub-questions of 3 to 5 marks will be asked.
- 4. Remaining questions will be mixed in nature.(e.g. Suppose Q.2 has part (a) from module 3 then part (b) will be from any module other than module 3)
- 5. In question paper weightage of each module will be proportional to number of respective lecture hrs as mentioned in the syllabus.
- 6. The first module (Communication Theory) will carry 40 % weightage.

- 1. Communication in Organizations by Dalmar Fisher, Jaico Publishing House
- 2. Communication Skills by Meenakshi Raman & Sangeeta Sharma,
- 3. Oxford University Press.
- 4. Business Correspondence & Report-writing by R.C. Sharma& Krishna Mohan, Tata McGraw-Hill Education.
- 5. Effective Technical Communication by Ashraf Rizvi, Tata McGraw-Hill.
- 6. Technical Writing & Professional Communication for non-native speakers of English by Thomas N. Huckin & Leslie A. Olsen, McGraw –Hill.
- 7. Mastering Communication by Nicky Stanton, Palgrave Master Series
- 8. www.buisnesscommunicationskills.com
- 9. www.kcitraing.com
- 10. www.mindtools.com
- 11. Journal of Business Communication

Course Code	Course Name		hing Scho ntact Hou		Credits Assigned			
		Theory	Pract.	Tut.	Theory	TW/Pract	Tut.	Total
FEL201	Basic Workshop Practice - II		04			02		02

Course Code	Course Name	Examination Scheme							
			T	heory					
		Inter	nal Asso	essment	End	Term	Pract	Oral	Total
		Test1	Test2	Av of Test 1 & 2	Sem Exam	Work			
FEL201	Basic Workshop Practice - II					50			50

Detailed Syllabus is given in Basic Workshop Practice-I

Term work:

Term work shall consist of respective reports and jobs of the trades selected the distribution of marks for term work shall be as follows:

Laboratory work (Job and Journal) : 40 marks

Attendance (Practical and Theory) : 10 marks

The final certification and acceptance of term - work ensures the satisfactory performance of laboratory work.